

“LINE-by-LINE” SOFTWARE FOR SPECTRA SIMULATION

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DLS

LAB

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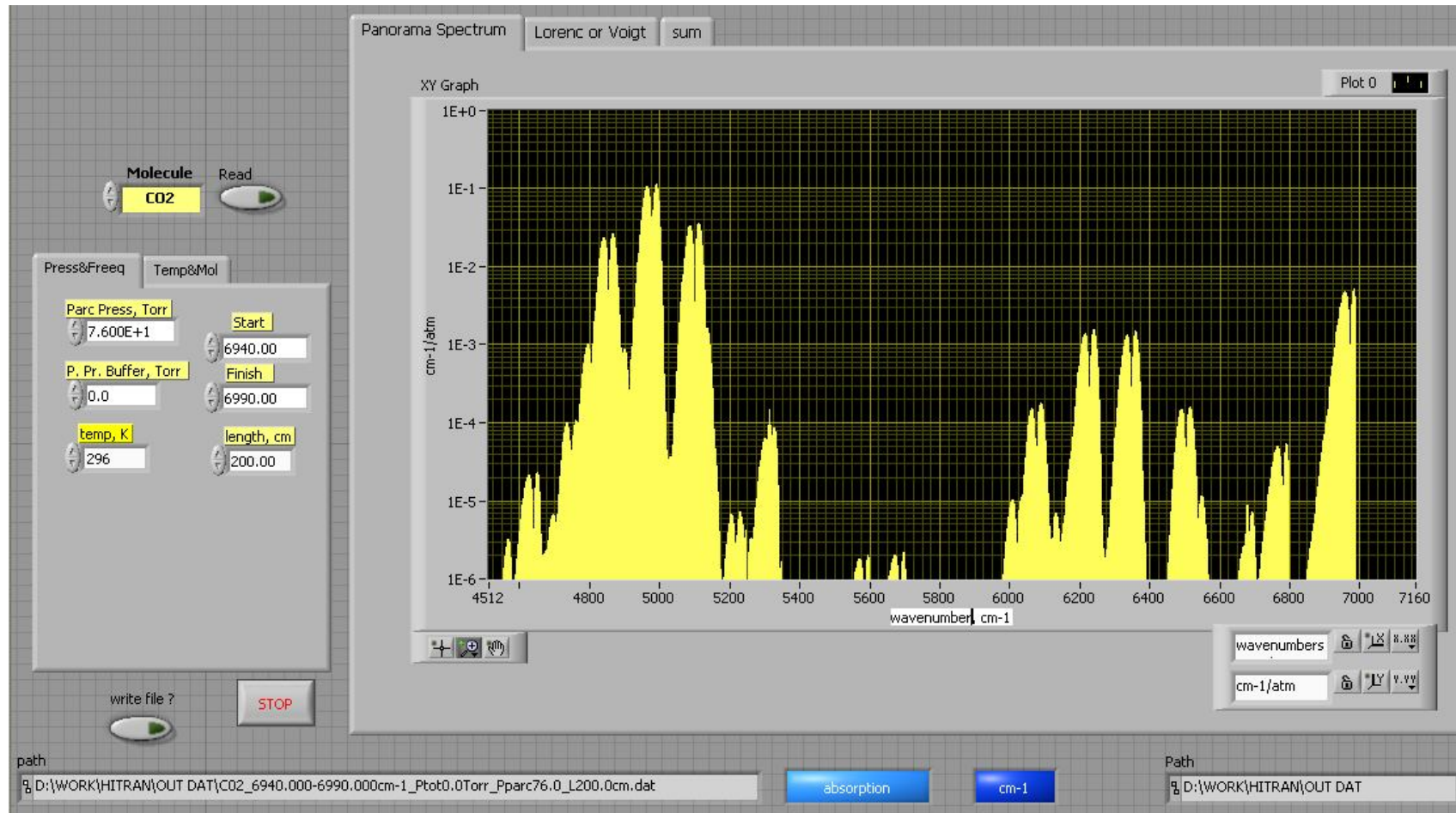
Introduction

TDLS analytical applications frequently need operative optimization of DL and analytical spectral line choice. Analysis has to be done taking into account absorption of other molecules in gas mixture under investigation. For atmosphere: H₂O, CO₂, CH₄, O₂, etc. To solve this problem, software “***Line-by-line***” was developed based on “LabView 8.6”, spectral database “HITRAN-2008” [1], [2] and Voigt line shape. Additional software option is recorded TDLS spectra comparison with simulation to identify recorded lines.

[1] <http://www.cfa.harvard.edu/HITRAN/>

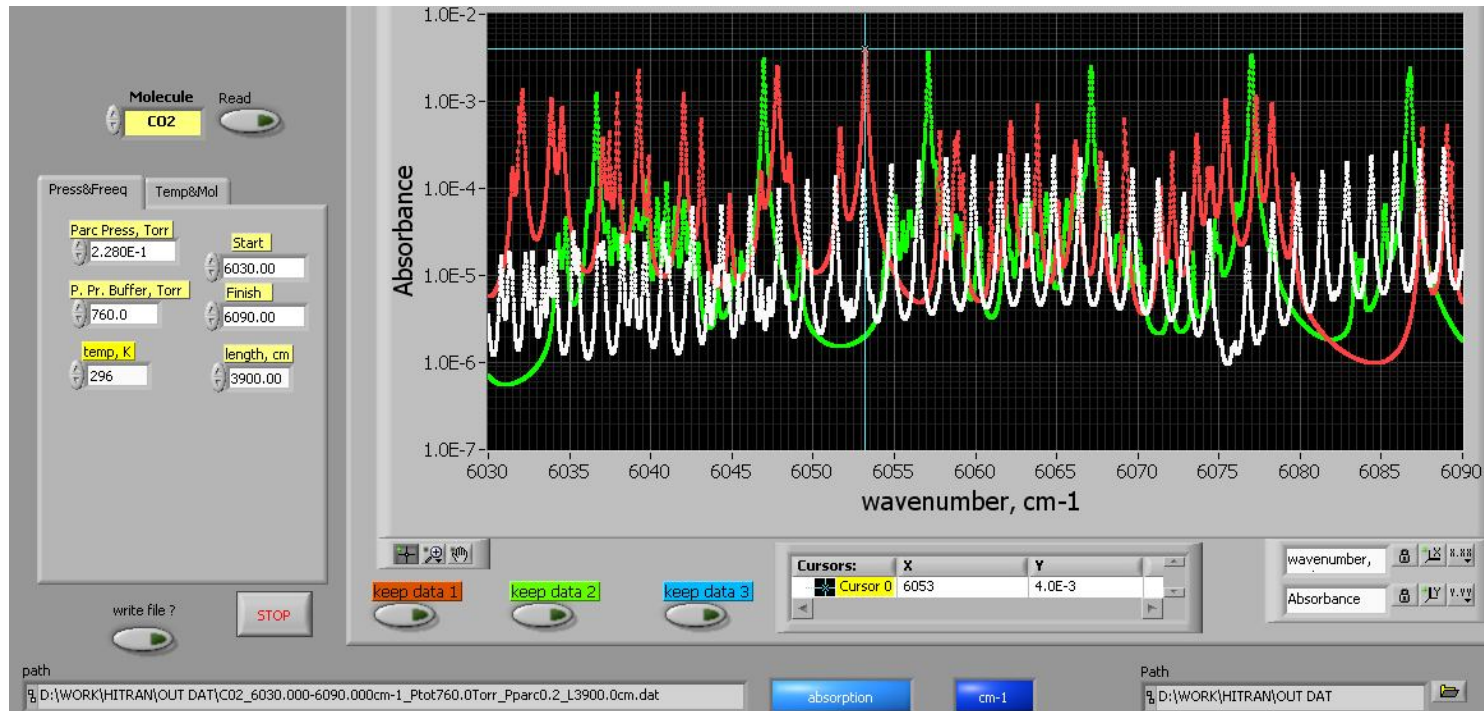
[2] L.S. Rothman, I.E. Gordon, A. Barbe, D.Chris Benner, P.F. Bernath, M. Birk, V. Boudon, L.R. Brown, A. Campargue, J.-P. Champion, K. Chance, L.H. Coudert, V. Dana, V.M. Devi, S. Fally, J.-M. Flaud, R.R. Gamache, A. Goldmanm, D. Jacquemart, I. Kleiner, N. Lacome, W.J. Lafferty, J.-Y. Mandin, S.T. Massie, S.N. Mikhailenko, C.E. Miller, N. Moazzen-Ahmadi, O.V. Naumenko, A.V. Nikitin, J. Orphal, V.I. Perevalov, A. Perrin, A. Predoi-Cross, C.P. Rinsland, M. Rotger, M.Simeckova, M.A.H. Smith, K. Sung, S.A. Tashkun, J. Tennyson, R.A. Toth, A.C. Vandaele, J. Vander Auwera, “The HITRAN 2008 molecular spectroscopic database”, JQRST, 110, pages 533-572 (2009).

HITRAN data



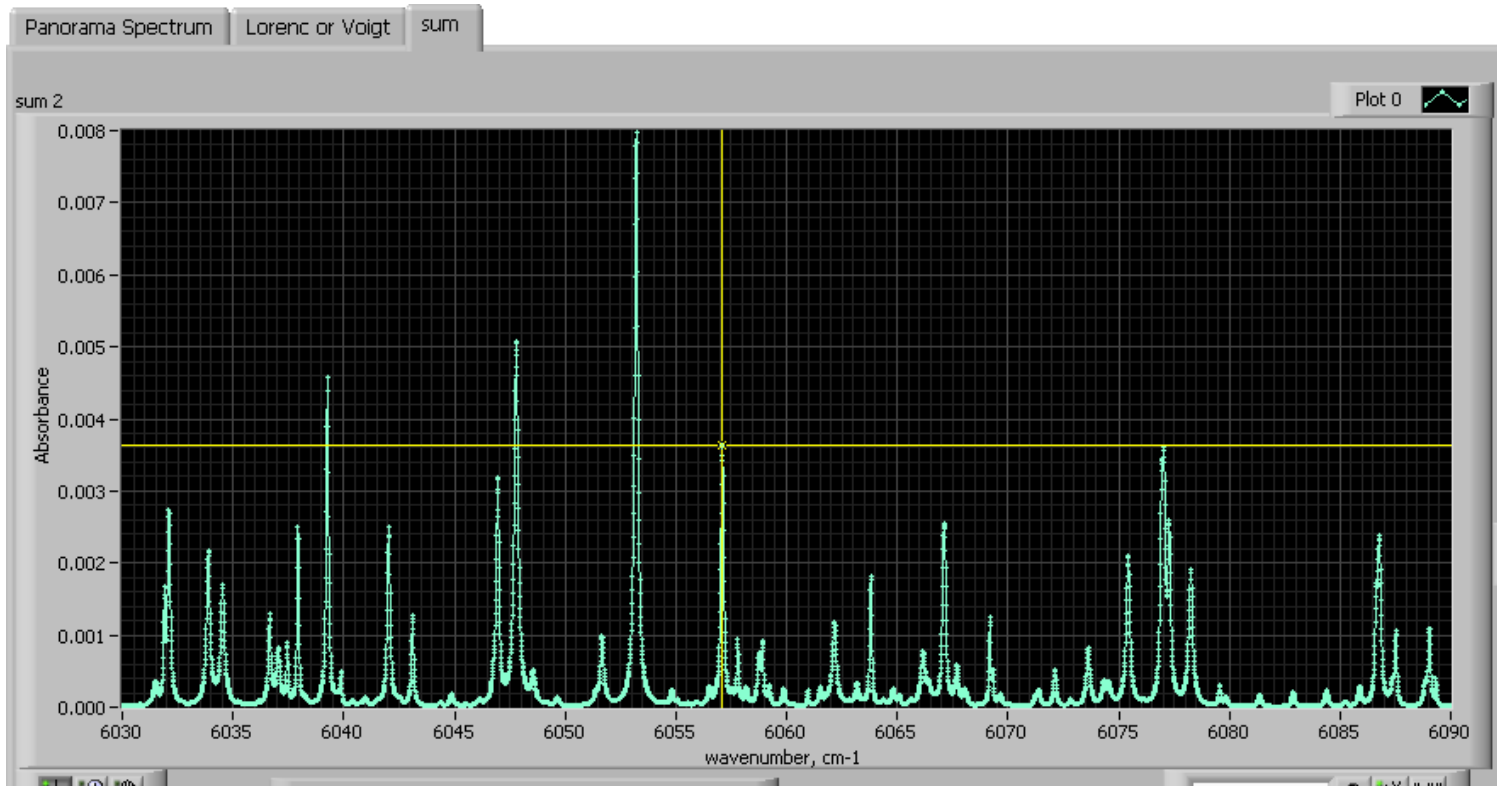
Preliminary data review

Software interface



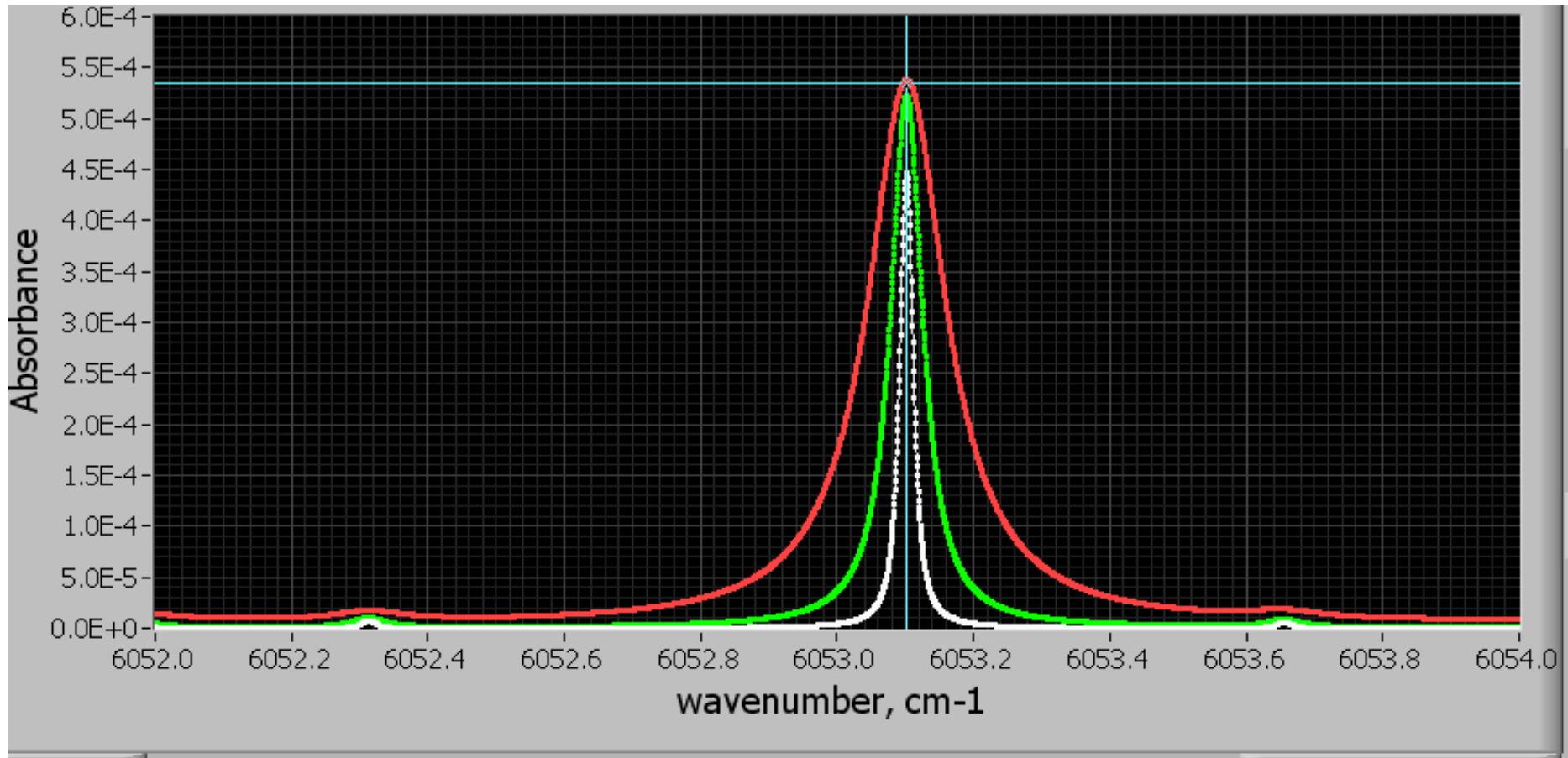
User can select molecules from HITRAN 2008, their partial pressures, total gas mixture pressure, temperature, optical length, and spectral range of interest. There are options to present result as absorbance or transmission vs wave number or wave length. Data calculated can be stored in computer. In present case CO₂ (white), H₂O (green), and CH₄ (red) were considered. Cursors (blue) can determine line position and its absorbance.

Complex gas mixture spectrum



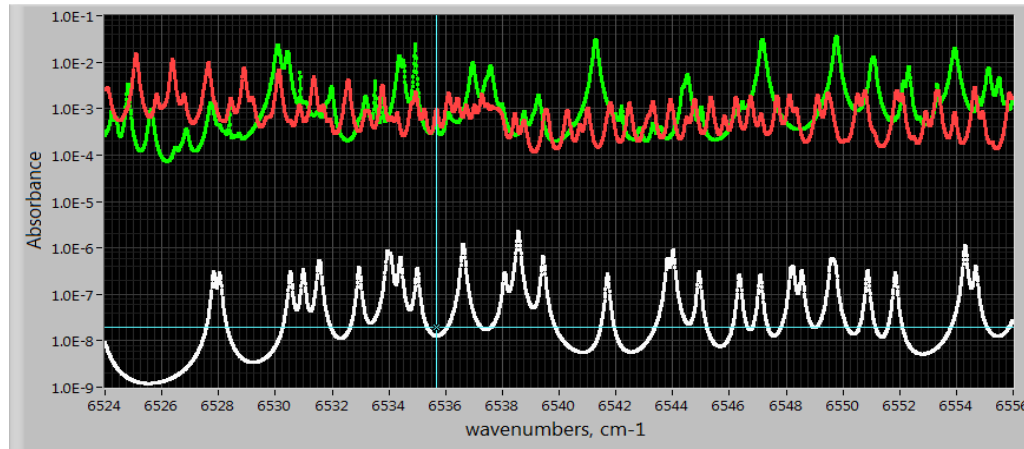
As option total absorbance of complex gas mixture under consideration (similar as on previous slide) and its derivatives can be calculated to be compared with data obtained by TDLS.

Pressure dependence

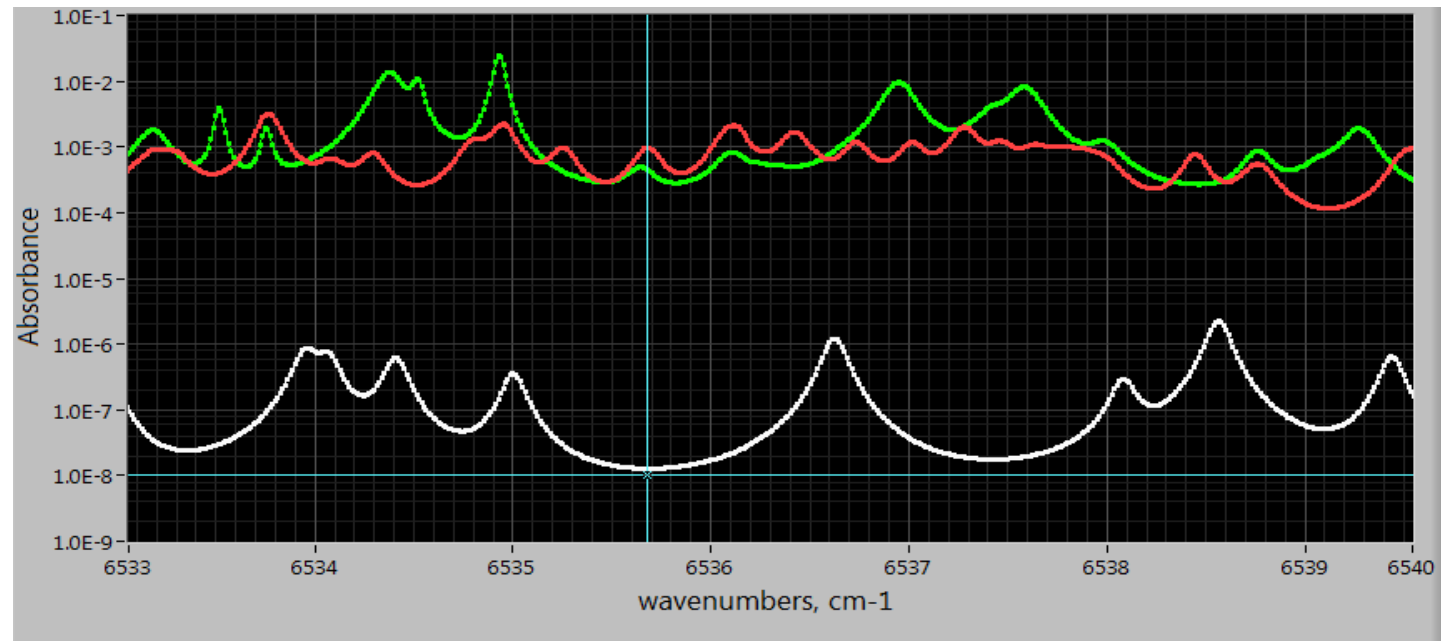


CO₂:air = 1:1000 absorbance spectrum simulation: L = 39 m, P = 760 (red), 300 (green), 100 (white) Torr.

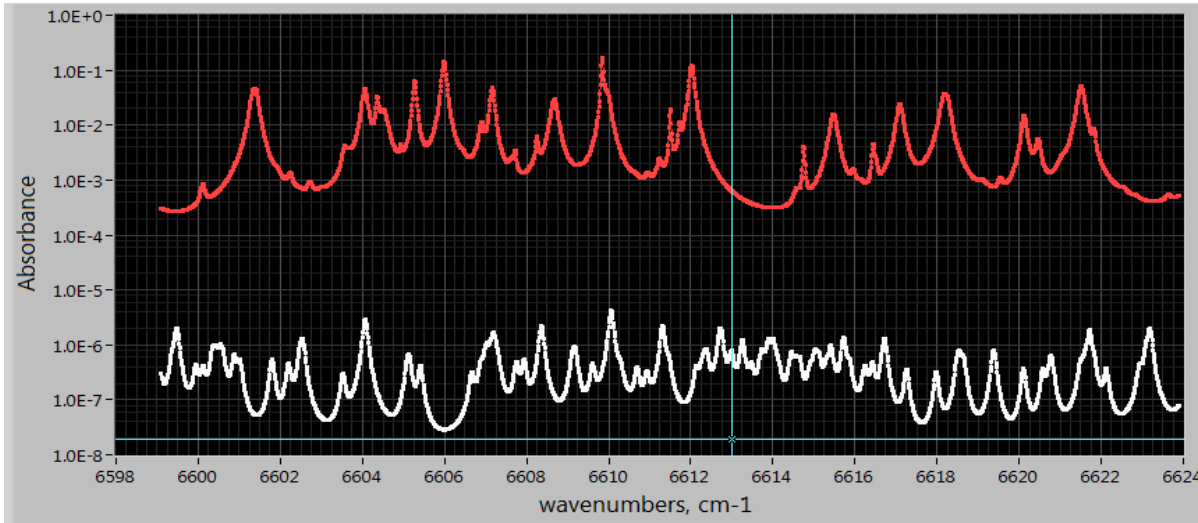
NH₃ detection in human breath in 1.53 μ



CH₄ =10.0 ppm (white),
H₂O =2 % (green),
CO₂ =3% (red) ,L=39 m,
P=1 atm, T=296 °K

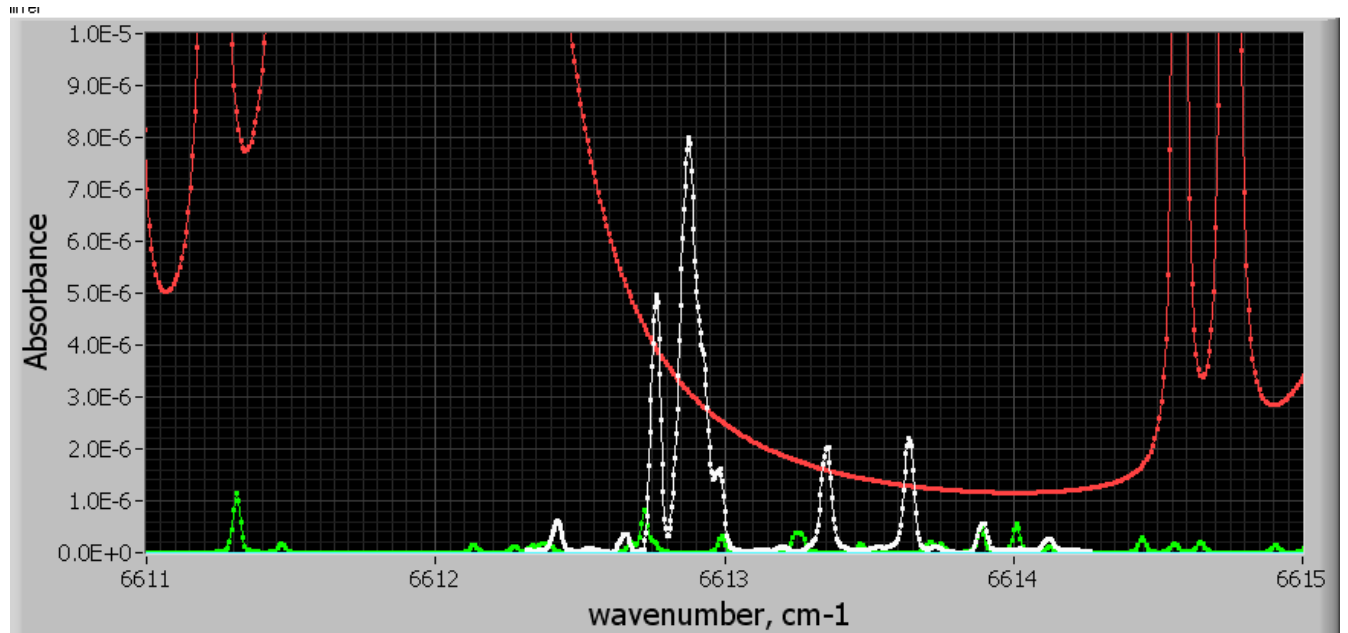


NH₃ detection in human breath in 1.51 μ

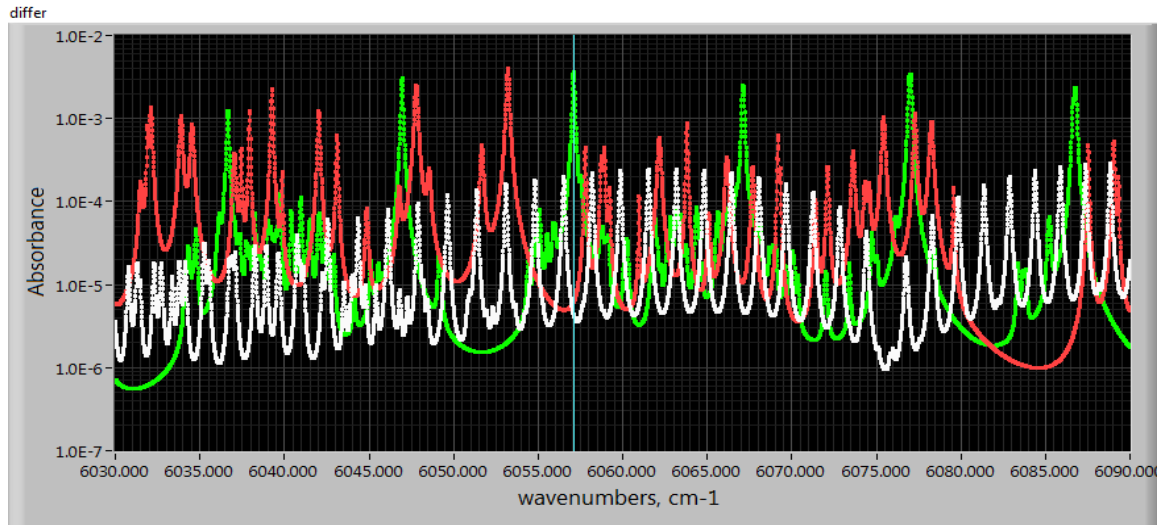


CH₄ =10.0 ppm (white),
H₂O =2 % (red),
L=39 m, P=1 atm,
T=296 °K

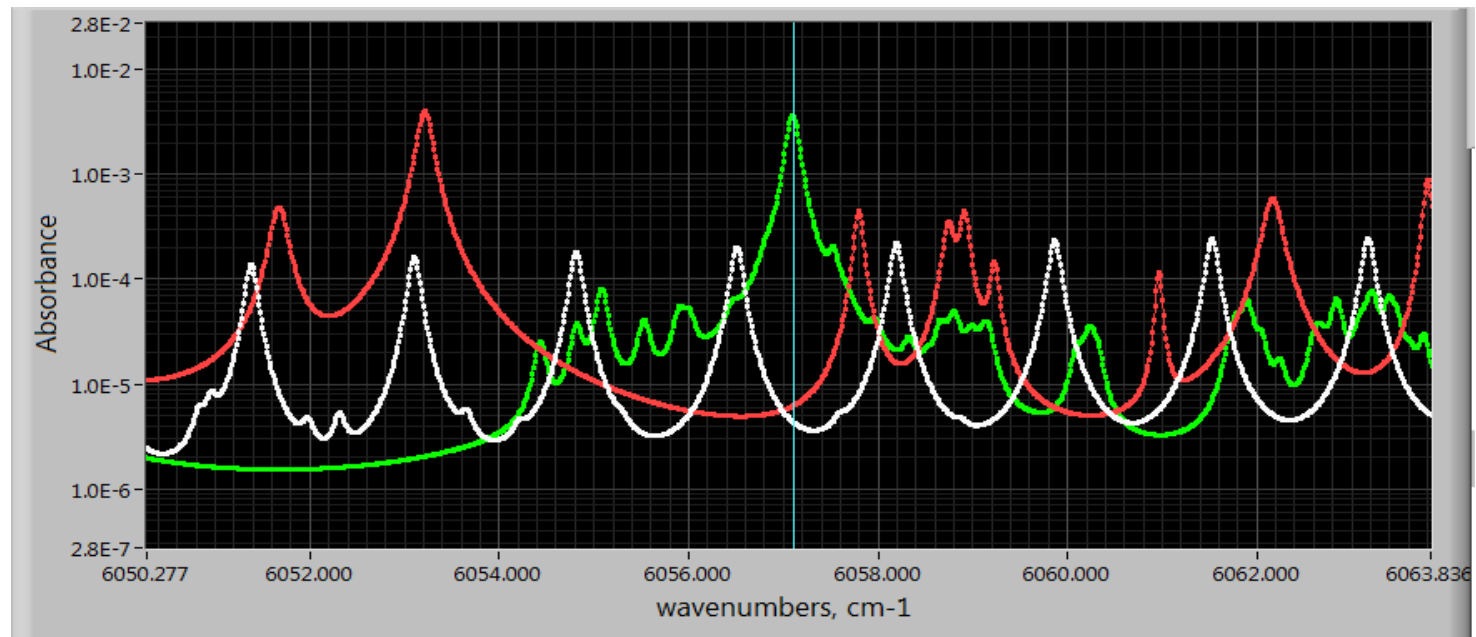
TDLDS spectrum of
NH₃=1.5 ppb (white)
CH₄ =10.0 ppm (green),
H₂O =2 % (red),
L=39 m, P=0.1 atm,
T=296 °K



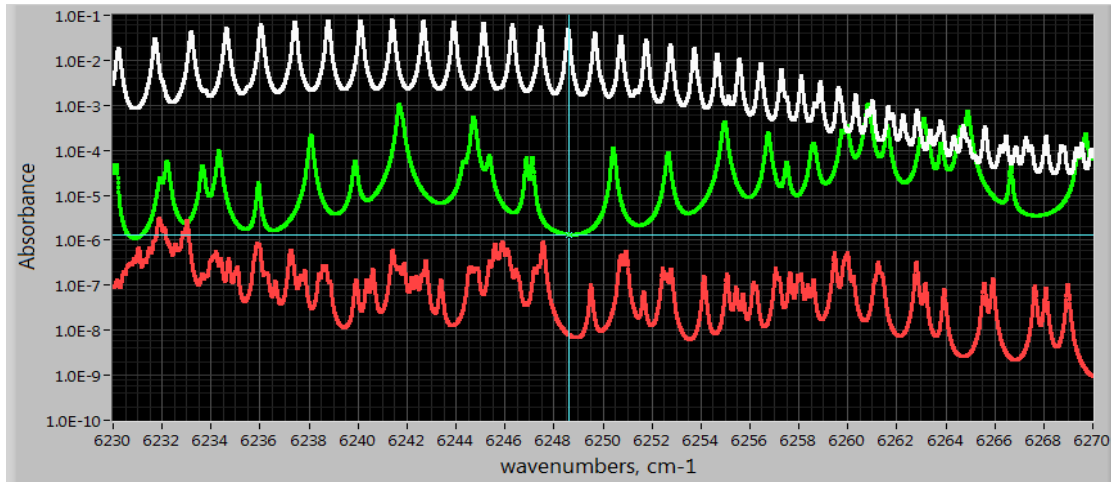
Trace CH₄ detection in 1.65 μ



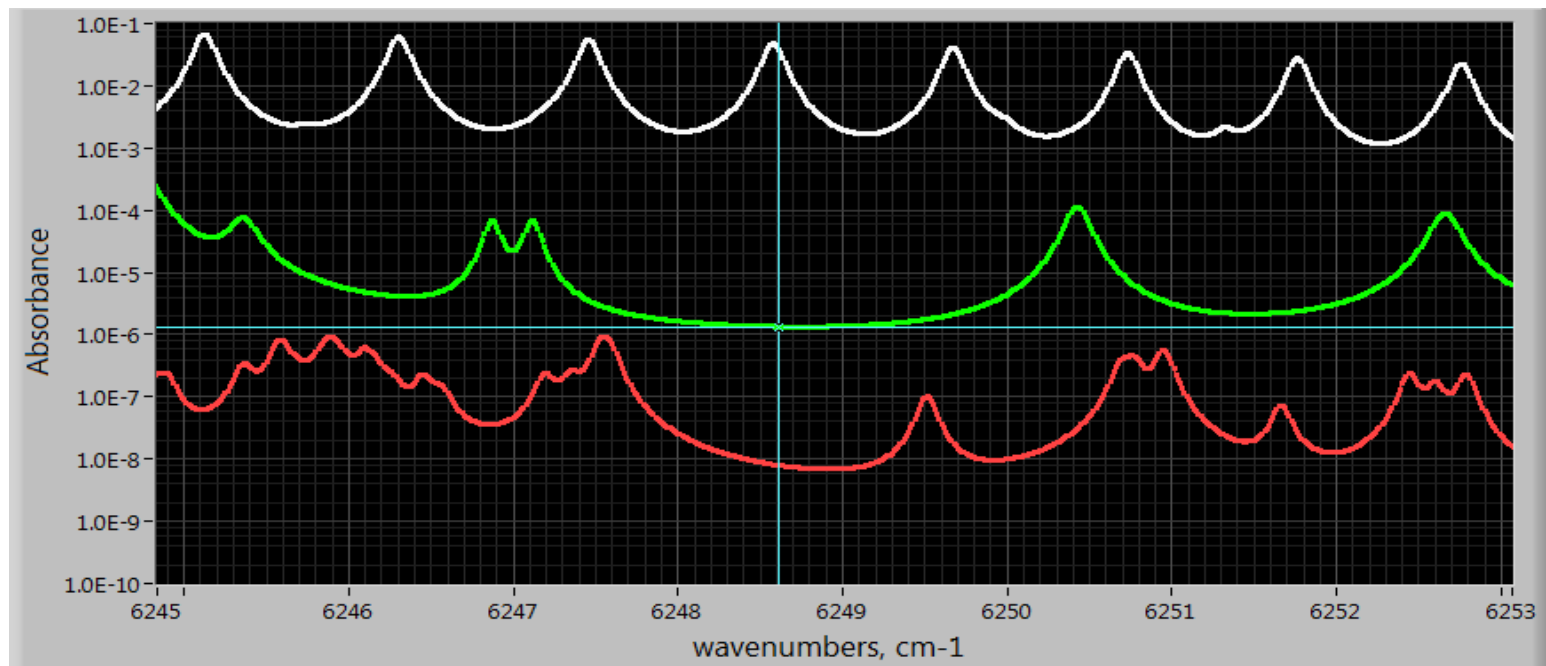
CH₄ = 2.0 ppm (green),
CO₂ = 300 ppm (white),
H₂O = 1 % (red), L=39 m,
P=1 atm, T=296 °K



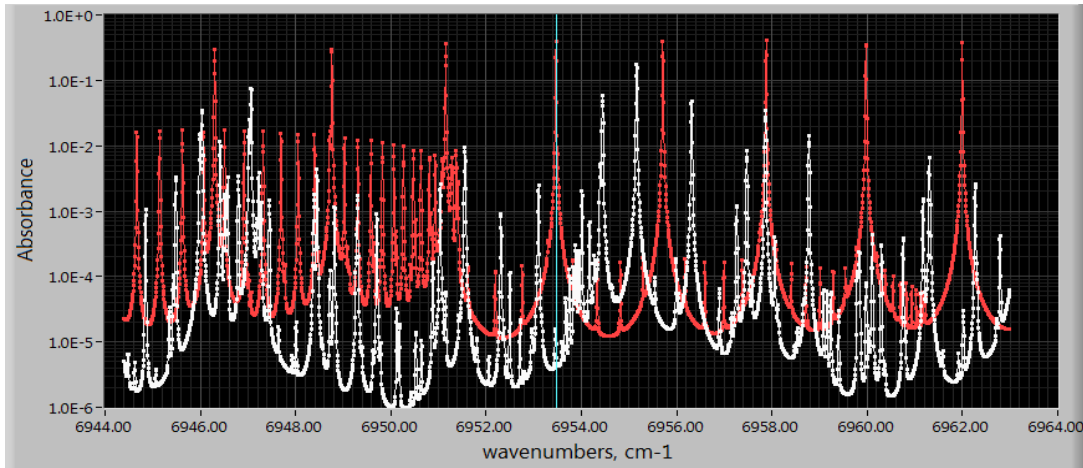
Trace CO₂ detection in 1.60 μ



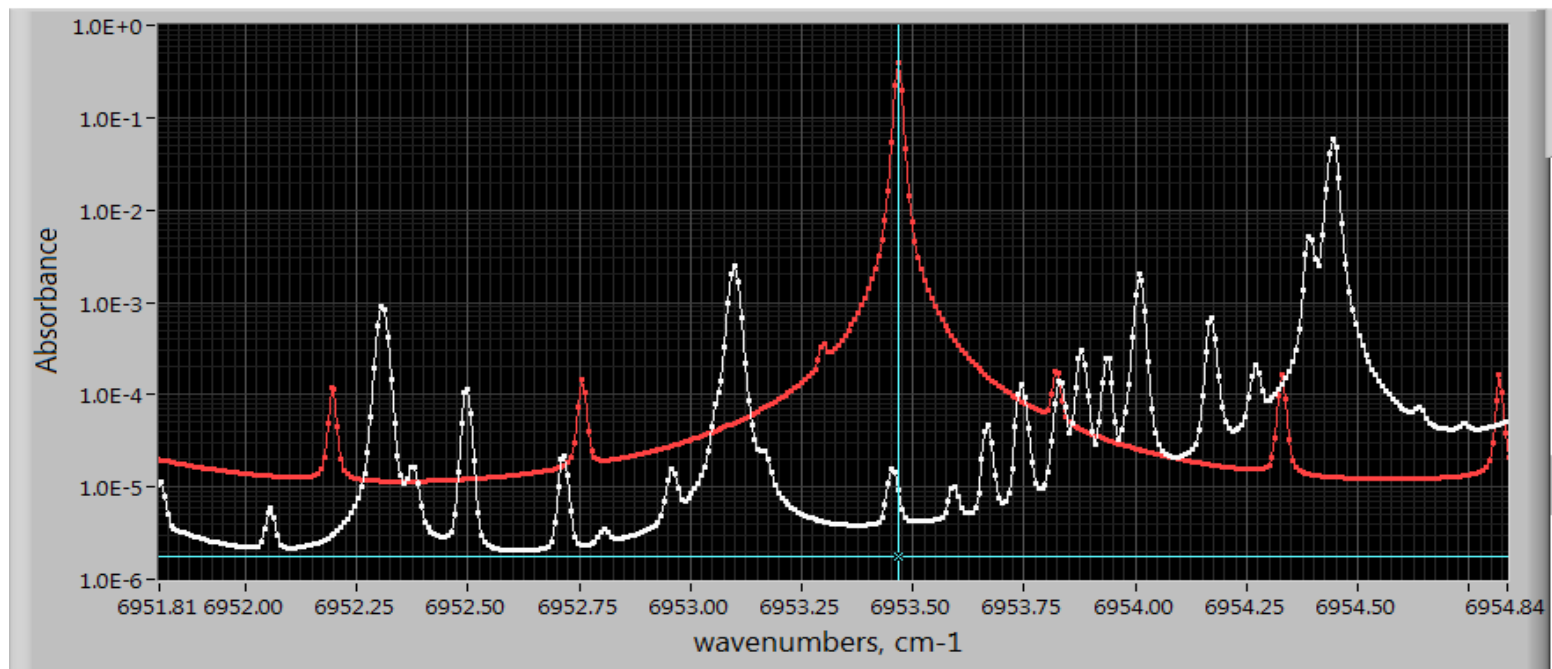
CO₂ = 300 ppm (white),
H₂O = 1 % (green), CH₄
= 2.0 ppm (red), L=39 m,
P=1 atm, T=296 °K



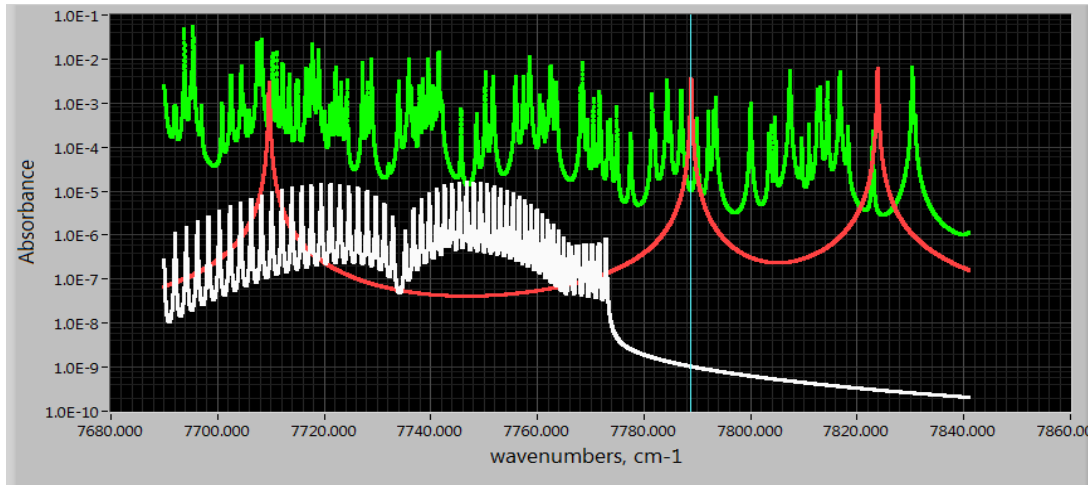
Precision measurement of contour line CO₂ in 1.44 μ



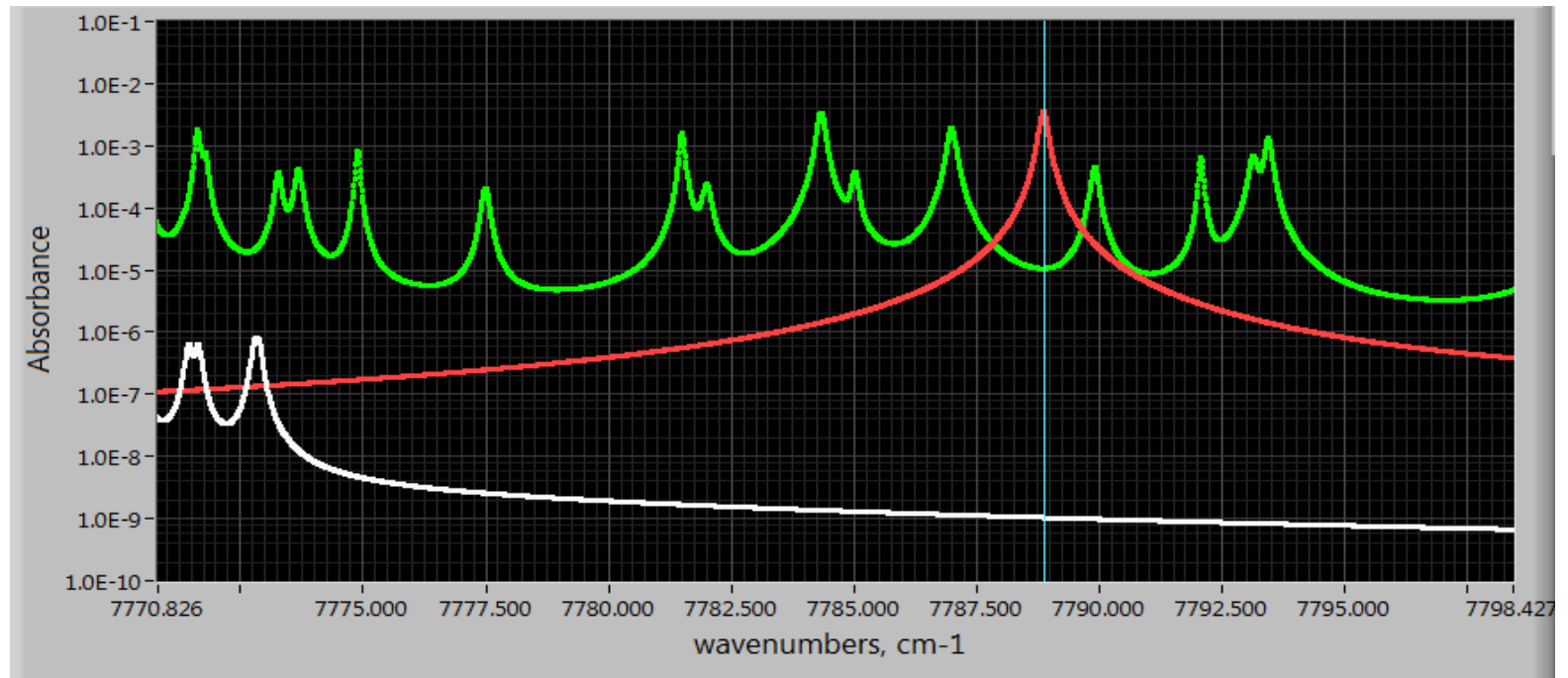
CO₂ = 2.63 kPa (red), H₂O
= 0.065 kPa (white) in 2.63 kPa
N₂, L = 200 cm, T = 296 °K



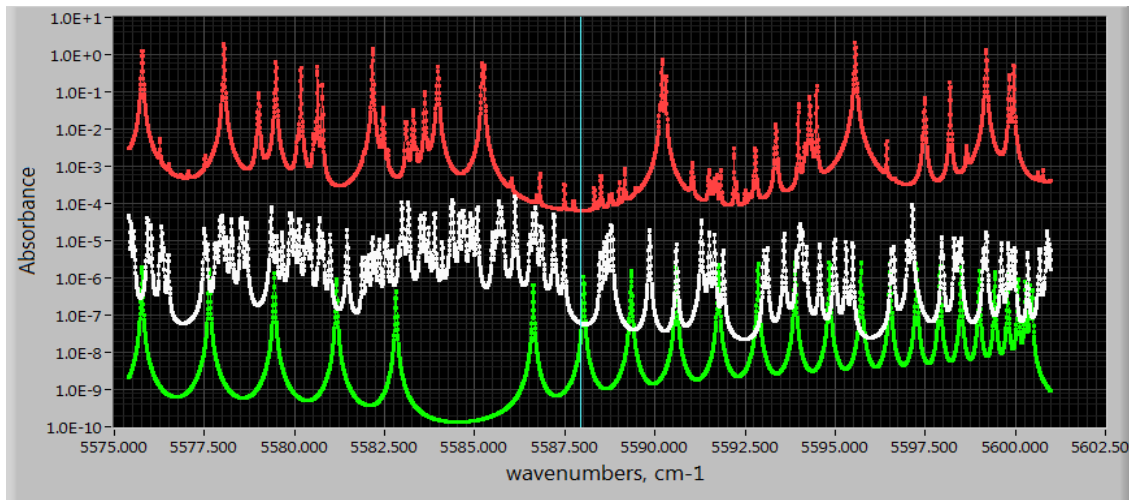
HF detection in 1.3 μ



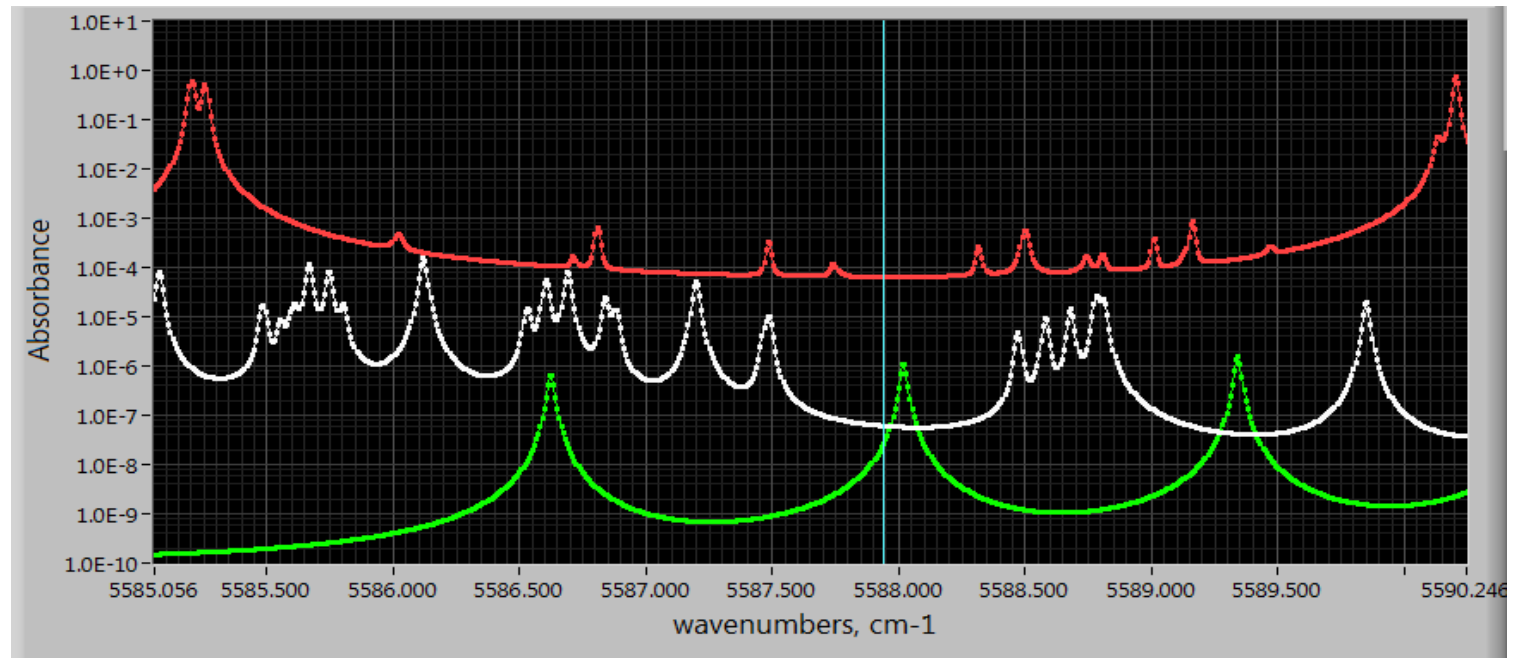
HF = 200 ppp (red), CO₂ = 300 ppm (white), H₂O = 1 % (green), L=39 m, P=1 atm, T=296 °K



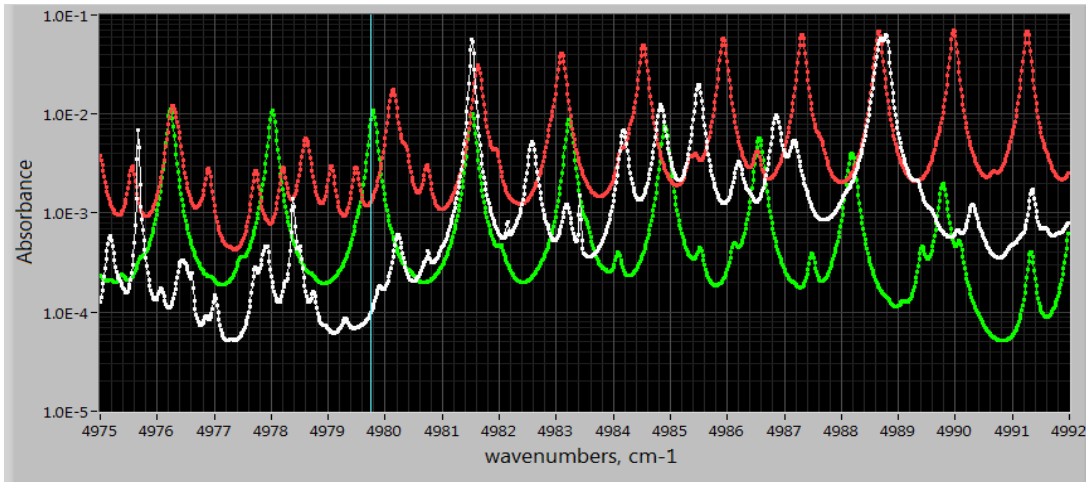
H₂CO detection in 1.79 μ



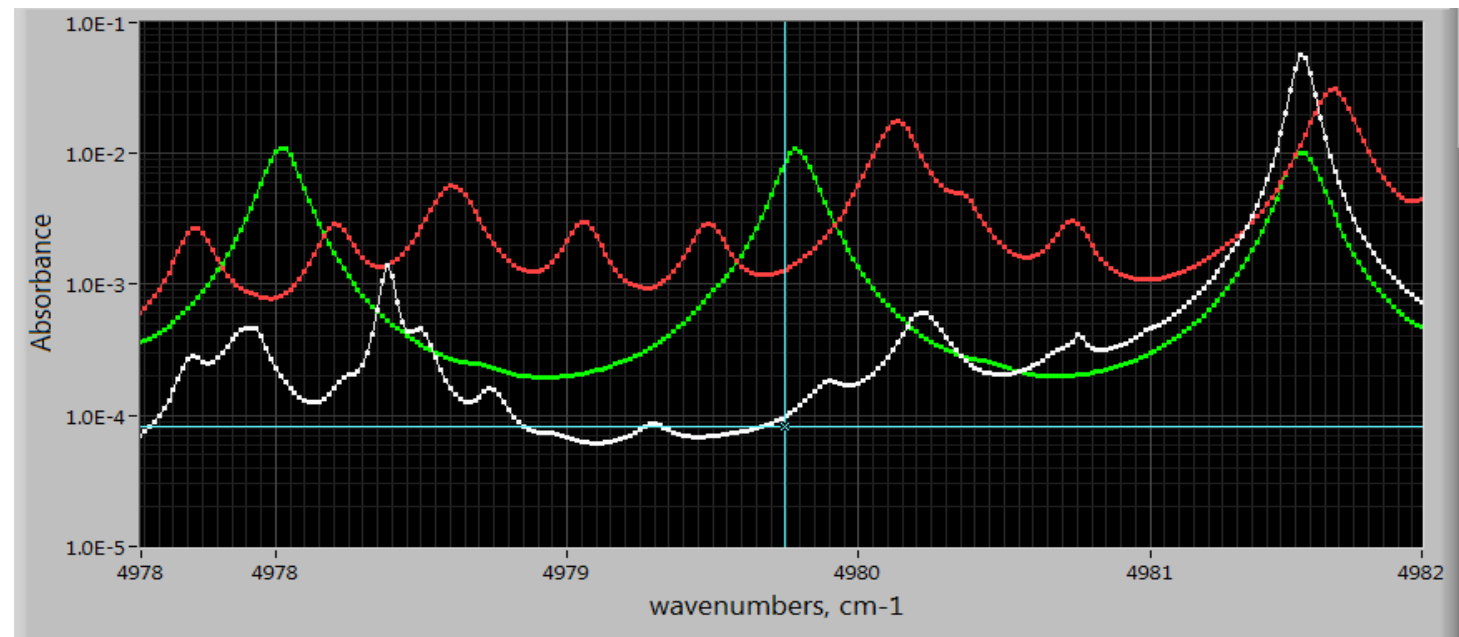
Analytical region for detection H₂CO. H₂O=1% (red), CH₄=2 ppm (white), CO₂=300 ppm (green)



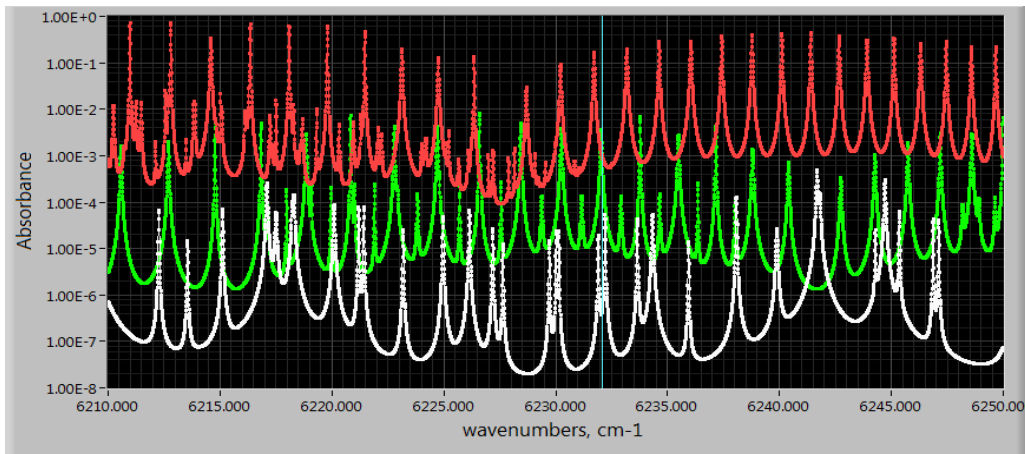
$^{13}\text{CO}_2 / ^{12}\text{CO}_2$ -ratio measurements in 2.0μ



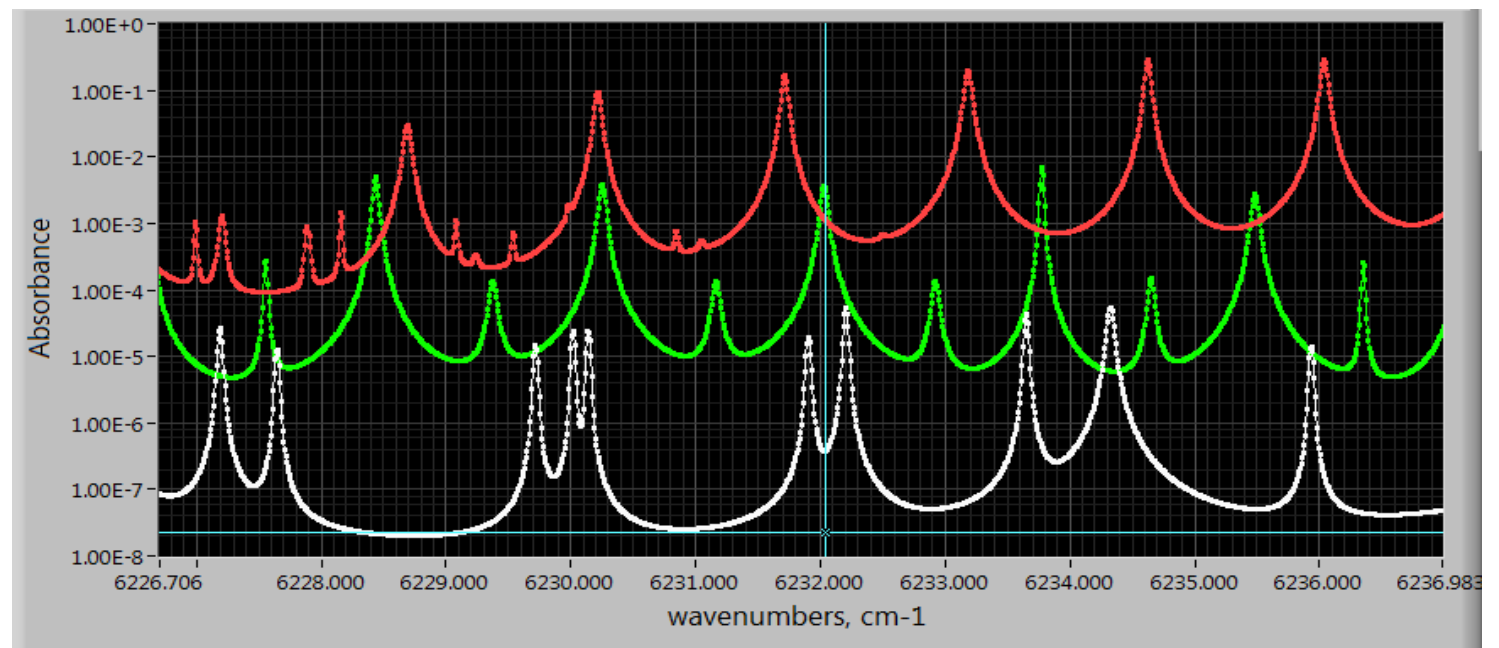
$^{12}\text{CO}_2 = 300 \text{ ppm}$ (red), $^{13}\text{CO}_2 = 30 \text{ ppm}$ (green), $\text{H}_2\text{O} = 1\%$ (white), $L=15 \text{ m}$, $P=1 \text{ atm}$.



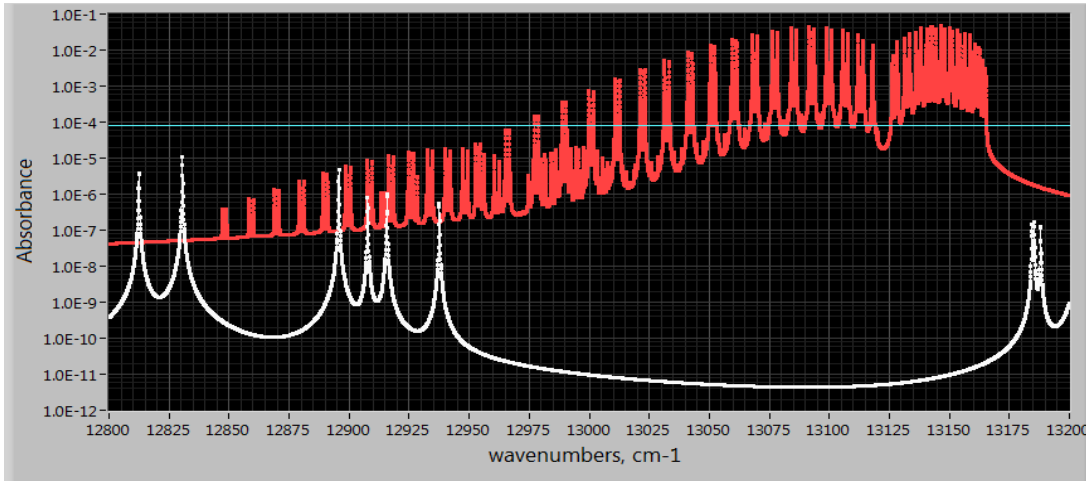
$^{13}\text{CO}_2 / ^{12}\text{CO}_2$ -ratio measurements in 1.6μ



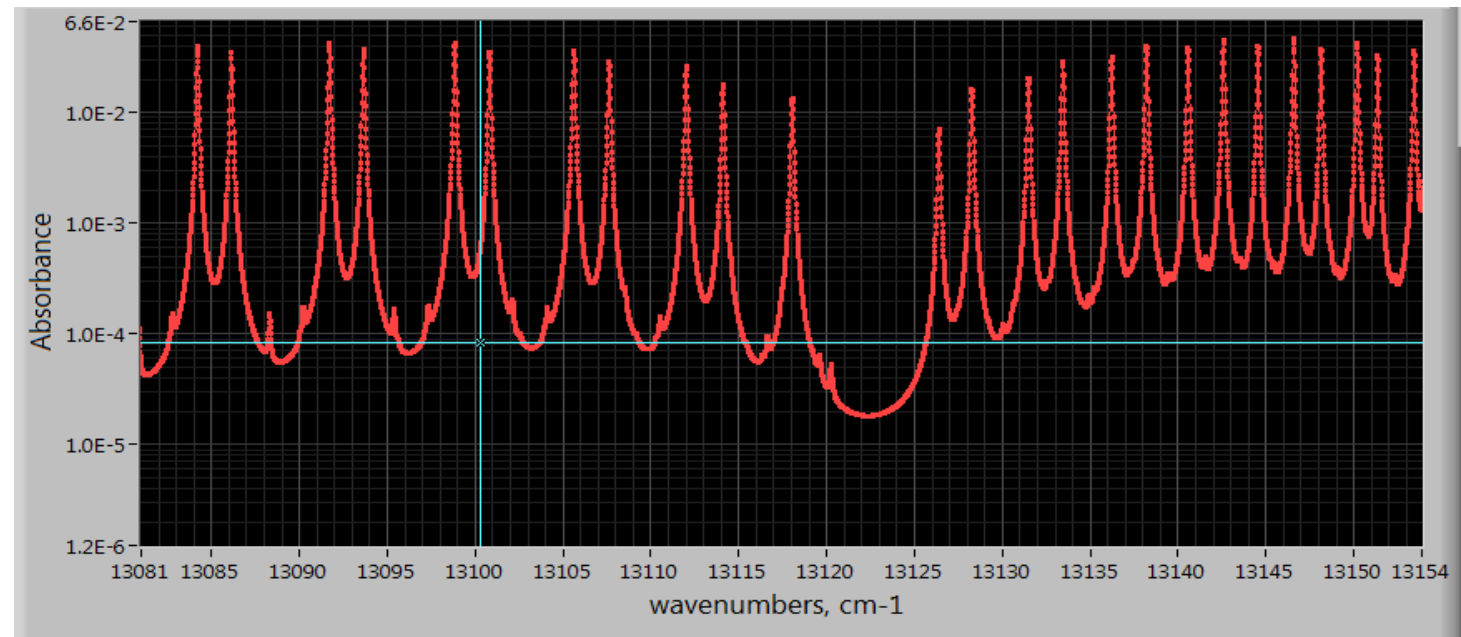
$^{12}\text{CO}_2 = 4\%$ (red), $^{13}\text{CO}_2 = 0.4\%$
(green), $\text{H}_2\text{O} = 3\%$ (white),
 $L=39 \text{ m}$, $P=0.13 \text{ atm}$.



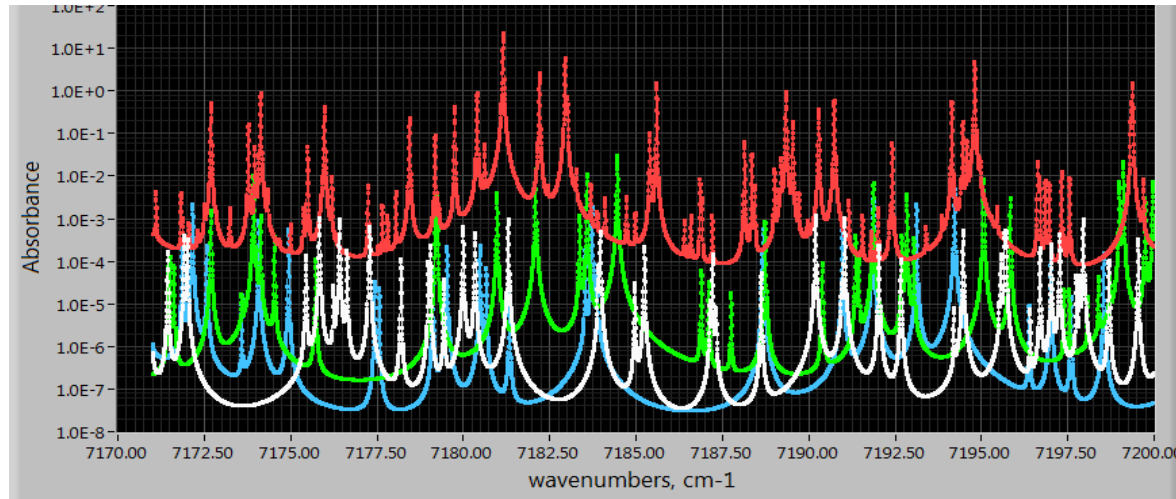
O₂ detection in 0.76 μ



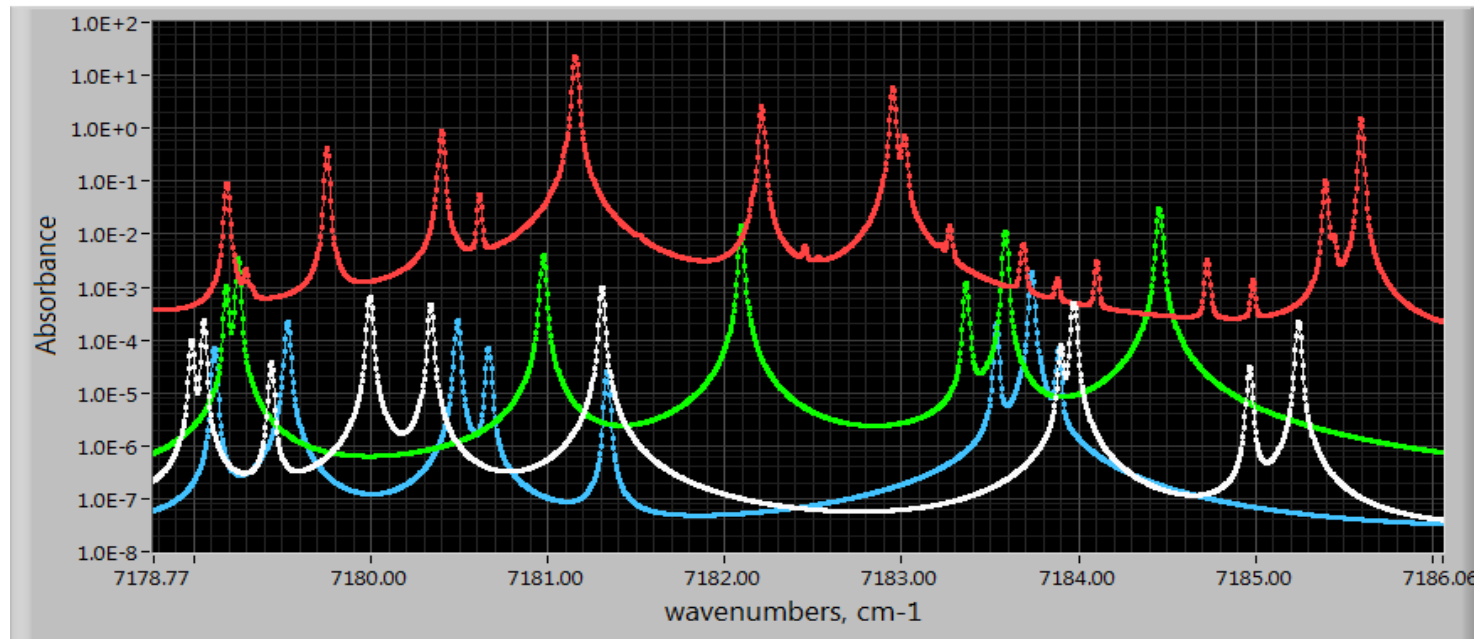
O₂ =20 % (red) , H₂O =1 %
(white), L=2 m, P=1 atm.



Accurate H₂O isotope analysis in 1.39 μ



H₂¹⁶O - red , H₂¹⁸O - green,
H₂¹⁷O – blue,
HDO – white.
P_{H₂O} =7 Torr, L=200 cm



Conclusion

Software “Line-by-line” was developed to simulate spectra of different molecules in near and mid IR spectral ranges where commercially available diode lasers operate. Simulation takes less than 1 min. Some examples are presented.

Simulation was performed for spectral ranges of operation for more than 50 NEL, Anritsu, Vertilas, Laser Components, and Nano Plus DLs available at GPI. Simulation results are used to select analytical line for particular DL for different applications. Some examples will be presented in poster.

Now similar simulation and analysis are performed before DL ordering.

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